

ASSIGNMENT OF THE ^{13}C NMR SPECTRUM OF QUINOVIC ACID

GHULAM A. MIANA and HASSAN M. G. AL-HAZIMI

Department of Chemistry, College of Science, King Saud University, Riyadh, Saudi Arabia

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Abstract—The ^{13}C NMR spectra of the triterpene quinovic acid and its methyl ester have been assigned. The structures of azizic and manevalic acids should be revised.

The ^{13}C NMR spectra of triterpene acids of the urs-12-ene (ursolic acid, 1) and olean-12-ene (oleanolic acid, 2) types have been well documented. Doddrell *et al.* [1] reported the stereochemical dependence of ^{13}C chemical shifts in the two series, and assigned chemical shifts for C-12, C-13 and for hydroxylated carbons. They found that ^{13}C NMR spectroscopy is the most precise tool for the purpose of distinguishing between the two types, which is not easy otherwise. Tori and co-workers [2, 3] gave complete assignments of the ^{13}C signals of several urs-12-enes and olean-12-enes.

^{13}C NMR spectra can be used to distinguish between the two types by observing the chemical shift of C-12 and particularly that of C-13. In general, C-12 is deshielded (-2 ppm) and C-13 shielded (-5 ppm) in the urs-12-ene series and appears at 140 ppm, whereas the latter resonates at 144-145 ppm in the olean-12-ene series. The difference between the two values has been rationalized by the presence of a 19β (equatorial) methyl group which is in close proximity to the double bond (γ and γ' to C-13 and C-12, respectively) in the urs-12-ene series, thus effecting sterically the chemical shifts of these carbons. Introduction of a 19α (axial) hydroxyl group was found to have a minor effect on the chemical shifts of C-12 and C-13, but the 19β -hydroxyl group has the same effect as the methyl group. The presence of a carboxyl group at C-14 (γ and γ' to C-13 and C-12, respectively) has a marked effect on the chemical shift. Thus the chemical shifts of C-12 and C-13 in 3 in the urs-12-ene series are at 128.5 and 133.5 ppm, respectively [1], while they are close to 126 and 137 ppm for 4, 5 and 6 in the olean-12-ene series [4].

We report herein the ^{13}C NMR spectra of quinovic acid (7) and dimethyl quinovate (8); this is the first report giving the complete assignment of a triterpene acid of the ursolic type bearing a $-\text{COOH}$ at C-14. Assignment of the carbon chemical shifts in the ^{13}C NMR spectra was made by the use of the single frequency off-resonance decoupling technique [5, 6], application of known chemical shift rules due to hydroxyl and acetoxyl substituent groups, and steric effects [1, 5, 7, 8], as well as by comparison with the ^{13}C NMR spectral data of known ursene and oleanene derivatives [9, 10]. The chemical shifts of 7 (Table 1), especially those of C-12 and C-13, which resonate at 128.6 and 133.7 ppm, respectively, clearly corroborate the results and conclusions of Doddrell *et al.* [1].

Table 1. ^{13}C NMR chemical shifts of compounds 7 and 8

Carbon No.	7	8	Carbon No.	7	8
1	39.02	38.60	16	25.24	24.65
2	26.06	25.13	17	47.02	48.49
3	77.80	78.70	18	54.60	53.72
4	39.02	38.74	19	37.10	36.90
5	55.48	56.12	20	37.20	36.90
6	18.72	18.25	21	30.35	29.80
7	36.87	32.00	22	36.87	35.98
8	39.71	39.7	23	27.82	27.18
9	48.43	47.30	24	16.08	16.30
10	37.20	37.05	25	16.40	16.30
11	23.13	22.77	26	18.02	15.60
12	128.60	129.28	27	177.70	175.40
13	133.70	131.98	28	179.88	177.80
14	56.47	51.48	29	18.72	17.00
15	28.80	28.06	30	21.19	21.00
			COOMe		51.10, 51.40

Application of these results to azizic acid (9) and manevalic acid (10) [11] (Table 2) clearly shows that the carboxyl group cannot be present at C-14 because the

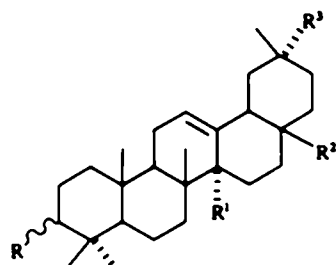
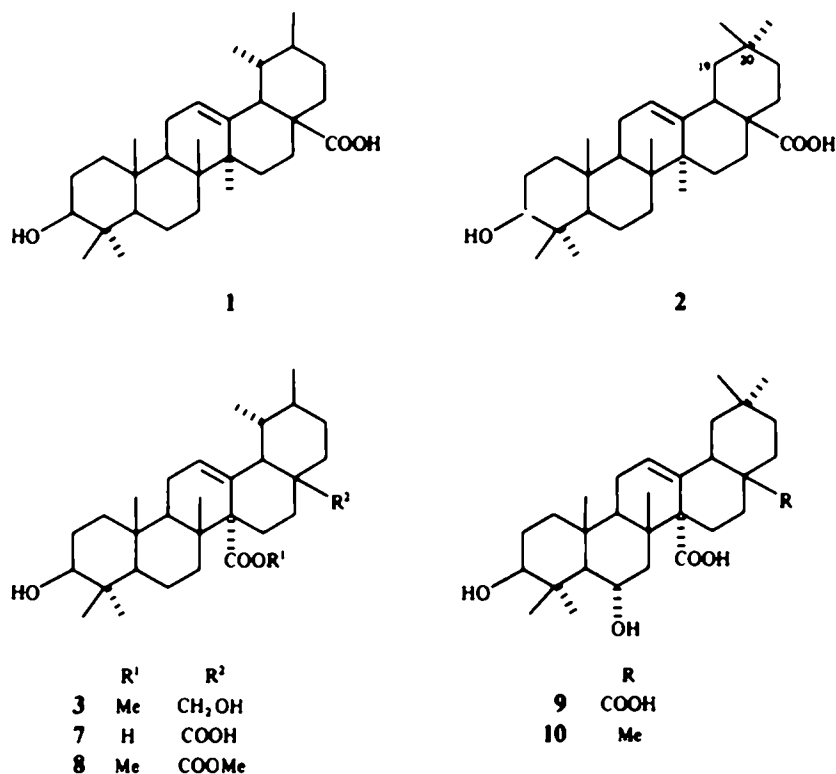
Table 2. Some ^{13}C NMR spectral data of compounds 4-6 and 9-13

Compound	C-12	C-13
4	126.0	137.6
5	126.8	136.1
6	125.5	137.5
9	123.0	143.1
10	122.3	143.8
11*	125.1	138.0
12†	122.8	144.7
13‡	122.8	142.8

* Data taken from ref. [12].

† Data taken from ref. [13].

‡ Data taken from ref. [14].



	R	R ¹	R ²	R ³
4	α -OH	COOH	Me	Me
5	α -OH	COOH	Me	CHO
6	α -OH	COOH	Me	COOH
11	β -OH	Me	COOMe	Me
2	α -OAc	Me	Me	COOMe
13	β -OH	Me	COOMe	COOMe

chemical shifts of C-12 and C-13 in both 9 and 10 have normal values of the oleanolic acid type as in 11-13 (Table 2) and show no shielding effect of a carboxyl group. In view of this finding, we believe that the structures of 9 and 10 should be revised.

EXPERIMENTAL

Quinovic acid was isolated from *Fagonia cretica* as described earlier in ref. [15], and its dimethyl ester was prepared in the same way. Quinovic acid was fully characterized by spectroscopic techniques by comparing the ¹H NMR and mass spectra with those described recently in the literature [16].

The ¹³C NMR spectra were obtained in CDCl₃ (pyridine-*d*₅) for quinovic acid at 25 MHz with TMS as internal standard.

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